

[Time: - 3 Hours]

[Marks: 70]

Please check whether you have got the right question paper.

- N.B: 1. All the questions are compulsory.
2. Figures to the right indicate full marks.

1. a. Convert the following: 02
1. 10 pound = _____ mg
 2. 400 minims = _____ ml
 3. 81 litres = _____ pints
 4. 20 grains = _____ mg
- b. Enlist the salient points to be considered as good pharmaceutical practices in compounding and dispensing laboratory. 02
- c. Calculate the dose of a drug for a 4 year child when the he adult dose of the same drug is 10 mg 01
- d. Give an account of liniments 02
- e. Give an account of suspension made by chemical reaction 02
- f. Compare and contrast o/w and w/o emulsion 02
- g. Define gels and classify gelling agents 02
- h. Enlist the advantages and disadvantages of capsules as a dosage form 02
2. a. How many ml of 1: 10000 w/v solution of preservative benzalkonium chloride can be made from 250 ml of 0.25 % solution 03
- b. Comment on the following prescription 04
- Rx
Sodium sulphate 30% w/w
Effervescent base qs
Dose: 1 tsp to be added to a tumblerful of water and consumed before breakfast
- OR**
- Classify powders. Describe the salient features of compounding and dispensing of tablet triturates.
- c. Enlist the various types of ointment bases. Include a note on compounding of ointments. 04
3. a. Classify creams. Write a note on preservation of creams 03
- b. Enlist the advantages of solution as a dosage form. Comment on the following prescription 04
- Rx
100 ml of zinc chloride and zinc sulphate mouthwash BPC
Zinc chloride 1% w/v
Zinc sulphate 2% w/v
Label: To be diluted with 20 times its volume of warm water before use

OR**TURN OVER**

Give an account of solutions taken orally

- c. Elaborate on compounding and dispensing of pastilles 04
4. a. What are emulsifying agents? Classify them giving examples 03
 b. Enlist the properties of a good suspension. Write a note on excipients used in the compounding of suspension. 04
- OR**
- Write a note on compounding and dispensing of suspensions containing precipitate forming liquids
- c. Draw and imaginary prescription and label its parts. Include a note on pricing of prescription 04
5. a. Find the amount of NaCl to be included in 100 ml of a 0.3% w/v solution of zinc sulphate so that, on dilution with an equal quantity of water, it will be iso-osmotic with tissue fluids. 03
 Given:
 Freezing point of 1% w/v solution of zinc sulphate is -0.076°C
 Freezing point of 1% w/v solution of sodium chloride is -0.576°C
- b. Give a detailed account of disadvantages of cocoa butter as a suppository base 04
- OR**
- Discuss polyethylene glycol as suppository base
- c. What in incompatibility. Discuss physical incompatibility 04
6. a. Give the labelling instructions for **any 2** of the following dosage forms: 02
 1. Lotions
 2. Suppositories
 3. Throat paints
- b. Give the English translation of the following Latin terms or abbreviations: 02
 1. Guttae
 2. Dolore urgente
 3. Dexter
 4. b.i.d.
- c. How would you dispense proprietary medicines? 03
 d. In what proportions would you mix Tween 80 (HLB 15) and Span 80 (HLB 4.5) to obtain 50 g of an emulgent having a HLB of 8 02
 e. Enlist the steps involved in compounding of suppositories made with macrogol base 02